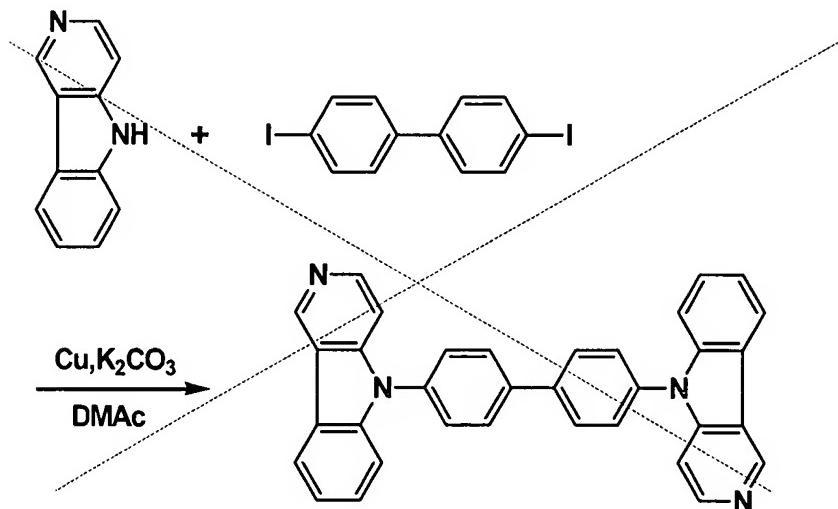
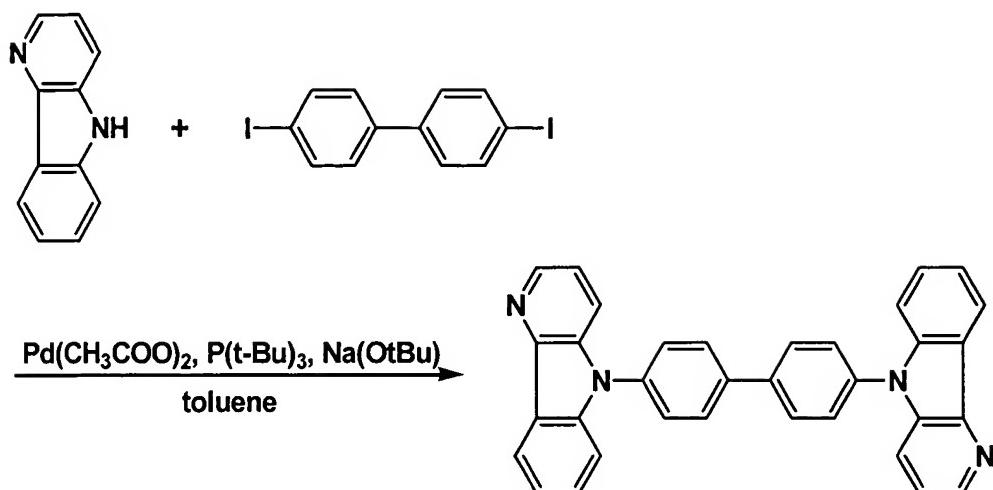


Amendments to the Specification:

Page 86, please replace the formula on the bottom of the page for Compound 74 as follows:

<<Preparation of Exemplified Compound 74>>





Replace the paragraph bridging pages 86-87 as follows:

0.32 g of palladium acetate and 1.17 g of tri-tert-butylphosphine were dissolved in 10 ml of anhydrous toluene and 50mg of sodium borohydride was further added. After the mixture was agitated for 10 minutes at room temperature, 5.00 g of δ -carboline, 5.87 g of 4,4'-diiodobiphenyl and 3.42 g of sodium-tert-butoxide were added and dispersed in 50 ml of anhydrous xylene, and agitated for 10 hours at the reflux

temperature under a nitrogen atmosphere. After standing the reaction mixture to cool, chloroform and water were added, and the organic layer was separated. The organic layer was washed with water and a saturated sodium chloride solution, and condensed under reduced pressure. The obtained residue was dissolved in tetrahydrofuran, treated with activated carbon, and recrystallized to obtain 5.0 g of colorless crystals of the exemplified compound 74.

Replace the paragraph bridging pages 89-90 as follows:

0.16 g of palladium acetate and 0.58 g of tri-tert-butyl phosphine were dissolved in 10 ml of anhydrous toluene, and 25mg of sodium borohydride was added. After the mixture was agitated for 10 minutes at ambient temperature, 2.00 g of δ -carboline, 3.20 g of an intermediate a and 1.37 g of sodium-tert-butoxide were added and dispersed in 50 ml of anhydrous xylene followed by stirring at reflux temperature for 10 hours under a nitrogen atmosphere. After standing the product to cool, chloroform and water were added to separate an organic layer. The organic layer

was washed by water and a saturated sodium chloride solution, and condensed under a reduced pressure. The residue was recrystallized in acetic acid, and 1.5 g of colorless crystal of the exemplified compound 144 was obtained.

Replace the paragraph bridging **pages 115-116** as follows:

A metal complex of an 8-quinolynol derivative such as aluminum tris(8-quinolynol) ~~(Alq)~~ (Alq₃), aluminum tris(5,7-dichloro-8-quinolynol), aluminum tris(5,7-dibromo-8-quinolynol), aluminum tris(2-methyl-8-quinolynol), aluminum tris(5-methyl-8-quinolynol), or zinc bis(8-quinolynol) ~~(Znq)~~ (Znq₂), and a metal complex formed by replacing the central metal of the foregoing complexes with another metal atom such as In, Mg, Cu, Ca, Sn, Ga or Pb, can be used as the electron transporting material. Furthermore, a metal free or metal-containing phthalocyanine, and a derivative thereof, in which the molecular terminal is replaced by a substituent such as an alkyl group or a sulfonic acid group, are also preferably used as the electron transporting material. The distyrylpyrazine derivative exemplified as a material for the

light emitting layer may preferably be employed as the electron transporting material. An inorganic semiconductor such as n-Si and n-SiC may also be used as the electron transporting material in a similar way as in the hole transporting layer.

Page 140, replace the second full paragraph as follows:

The results are given in the following table. ~~Each value of luminance, external quantum efficiency and storage stability given in the following table was expressed as a relative value when each value for Organic EL Element Nos. 3-6 was set as 100.~~

Page 142, replace the first full paragraph as follows:

The results are given in the following table. ~~Each value of luminance, external quantum given in the following table was expressed as a relative value when each value for Organic EL Element 4-6 was set as 100.~~

Replace paragraph bridging **pages 143-144** as follows:

The results are given in the following table. ~~Each value of measured luminance, external quantum given in the following table was expressed as a relative value when each value for Organic EL Element 5-6 was set as 100.~~

Page 145, replace last paragraph as follows:

The results are given in the following table. ~~Each value of measured luminance, external quantum given in the following table was expressed as a relative value when each value for Organic EL Element 6-6 was set as 100.~~

Page 147, replace second paragraph as follows:

The results are given in the following table. ~~Each value of measured luminance, external quantum efficiency given in the following table was expressed as a relative value when each value for Organic EL Element 7-6 was set as 100.~~

Page 148, replace last paragraph as follows:

The results are given in the following table. Each value of measured luminance, external quantum efficiency and 50 °C driving life given in the following table was expressed as a relative value when each the value for Organic EL Element 8-6 was set as 100.

Page 150, replace the second paragraph under the heading as follows:

The results are given in the following table. Each value of measured luminance, external quantum efficiency and initial life given in the following table was expressed as a relative value when each the value for Organic EL Element 9-6 was set as 100.

Page 151, replace the third paragraph under the heading of Example 10 as follows:

The results are given in the following table. ~~Each value of measured luminance and external quantum efficiency given in the following table was expressed as a relative value when each value for Organic EL Element 10-6 was set as 100.~~ Each driving voltage was obtained as a voltage difference between the voltage of each organic EL element and that of Organic EL Element No. 10-6.